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Three-Dimensional Network ZnO/BaFe₁₂O₁₉ Composite Thick Films and their Microwave Absorption Properties

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Abstract

ZnO/BaFe₁₂O₁₉ composite thick films with four different ZnO/BaFe₁₂O₁₉ mass fractions were synthesized with the tape-casting method in the presence of plate-like BaFe₁₂O₁₉ grains and sphere-like ZnO grains. Their phase composition, morphology and magnetic properties were analyzed by means of XRD, SEM and VSM, respectively. The microwave absorption properties were also investigated in the frequency range of 2–18 GHz, and the results show that the ZnO/BaFe₁₂O₁₉ composite thick films have multiple microwave attenuation peaks and their microwave absorption properties can be easily tuned by varying the mass ratio of BaFe₁₂O₁₉/ZnO. When the mass ratio of BaFe₁₂O₁₉ to ZnO is 15:85, the composite thick film exhibits a minimum RL of -48.6 dB at 17.2 GHz with a thickness of 4.5 mm. The electromagnetic performance of the ZnO/BaFe₁₂O₁₉ composite thick films could be attributed to the effective complementarities between the dielectric loss and the magnetic loss.

Keywords: Tape casting, impedance matching, microwave absorption

I. Introduction

Microwave-absorbing materials have attracted growing attention because of their important applications in protecting the environment from the radiation of telecommunication equipment, and in some specific aircraft, communication devices, etc. ^{1–6}. ZnO is considered to be a promising microwave-absorbing material in civilian and military applications because of its light weight, low cost, simple synthesis, and strong absorption ability ^{7–11}. However, pure dielectric materials with high dielectric constants are harmful for impedance matching, which leads to strong reflection and weak absorption. Therefore, researchers are urged to find new types of electromagnetic-absorbing materials to deal with electromagnetic radiation.

The introduction of magnetic materials to form ZnObased composites is a good solution to make up for the deficiency of ZnO. Recently, an increasing number of reports have focused on synthesizing ZnO-based composites because of their unique properties with possible technological applications in the microwave absorption field ^{12, 13}; these include, for example, SrFe₁₂O₁₉/ZnO ¹⁴, ZnO/CoFe₂O₄ ¹⁵ and ZnAl₂O₄/ZnO ¹⁶. To the best of our knowledge, there are few reports on combining ZnO and BaFe₁₂O₁₉) to form composites. M-type Ba-hexaferrite (BaFe₁₂O₁₉) is a promising microwave-absorbing material owing to its axis-anisotropy behavior and magnetization value at the microwave frequencies ^{17–19}. Furthermore, its magnetic and dielectric properties could be modulated to satisfy different applications.

So far, various methods have been reported for synthe-

sizing ZnO-based composites, such as the in-situ hydrolysis method ¹⁴, sol-gel process ^{15, 16} and hydrothermal method ²⁰. However, the above methods have many disadvantages such as expensive instruments, harsh reaction conditions, toxic reagents and long reaction time. Therefore, it is necessary to explore a simple, fast and low-cost method for the synthesis of ZnO-based composites. The tape-casting method has been demonstrated to be one of the more simple and feasible methods ^{14–16, 20}. The tapecasting method can also be easily adapted to prepare multi-layer composites. In this paper, ZnO/BaFe₁₂O₁₉ composite thick films with different mass ratios of $BaFe_{12}O_{19}$ were prepared with the tape-casting method $^{21-24}$. The phase composition, morphology and microwave absorption properties of ZnO/BaFe12O19 composite thick films were also investigated in detail.

II. Experimental

(1) Materials

Barium carbonate (BaCO₃), iron sesquioxide (Fe₂O₃), sodium chloride (NaCl), barium chloride dihydrate (Ba-Cl₂·2H₂O), zinc nitrate hexahydrate (Zn(NO₃)₂·6H₂O) and sodium carbonate (NaCO₃), glycerol trioleate, polyvinyl butyral (PVB), dibutyl phthalate and anhydrous ethanol were purchased from Sinopharm Chemical Reagent Co., Ltd. All the chemical reagents used in this study were of analytical pure grade and used without further purification.

(2) Synthesis of $ZnO/BaFe_{12}O_{19}$ composite thick films

The plate-like $BaFe_{12}O_{19}$ powders were synthesized with a two-step molten salt method (MSM). Firstly, Ba-

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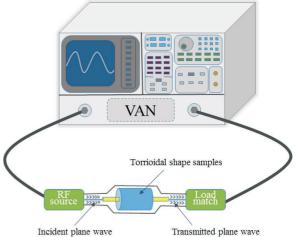
CO₃ and Fe₂O₃ were mixed according to the stoichiometric ratio of BaFe₁₂O₁₉. The mixed powders were calcined at 1000 °C for 2 h together with NaCl at the weight ratio of 1:2. The molten salt was washed with hot deionized water. Secondly, to coarsen the plate-like BaFe₁₂O₁₉ powders obtained from the MSM, the obtained powders were mixed with BaCl₂·2H₂O and Fe₂O₃. The mass ratio of BaCl₂·2H₂O and Fe₂O₃ was 2:1 and the mass ratio of Fe₂O₃ and BaFe₁₂O₁₉ ratio was 500:1²⁵. The mixtures were calcined at 1150 °C for 8 h to synthesize plate-like BaFe₁₂O₁₉ powders.

Sphere-like ZnO powders were synthesized with the sol-gel method. In a typical preparation procedure, 5 g $Zn(NO_3)_2 \cdot 6H_2O$ and 3 g NaCO₃ were dissolved in 30 mL deionized water, respectively. The NaCO₃ solution was slowly added to $Zn(NO_3)_2 \cdot 6H_2O$ solution and then the viscous gels were formed. The resulting product was collected by means of centrifugation, washed repeatedly with ethanol, and dried in vacuum at 80 °C for 8 h. Finally, the product was calcined in air at 500 °C for 2 h.

Composite thick films of $(1-x)ZnO/xBaFe_{12}O_{19}$ (x = 5 wt%, 15 wt%, 25 wt%, 35 wt%) were prepared with the tape-casting method. To prepare the homogeneous slurry, the ZnO powders and $BaFe_{12}O_{19}$ powders were mixed according to different mass ratios, together with the solvent (alcohol, 60 wt% of the powders, and butanone, 90 wt% of the powders), dispersant (glycerol trioleate, 2 wt% of the powders), binder (PVB, 5 wt% of the powders) and plasticizer (dibutyl phthalate, 3 wt% of the powders). Then the tape casting was carried out with the doctor blade height of 20 µm and at a speed of 30 cm/min. The resulting ZnO/BaFe₁₂O₁₉ composite thick films were dried at 100 °C for 10 h.

(3) Characterization

Phase analysis of the samples was performed with an Xray diffractometer (XRD, Rigaku D/MAX-2400, Japan) using CuK α radiation. The morphology of the samples was investigated with a field emission scanning electron microscopy (SEM, Hitachi S-4800, Japan). The magnetic properties were measured at room temperature on a vibrating sample magnetometer (VSM, Lakeshore-7410, Westerville, OH). The S parameters including S11, S12, S21 and S22 will be tested by a HP8720ES vector network analyzer using the coaxial-line method. Scheme 1 illustrates the schematic representation of typical electromagnetic parameters measurement setup. The testing samples were prepared by cutting a single layer and then pressing into toroidal-shaped samples (\emptyset out:7.0 mm, \emptyset in:3.04 mm). Afterward, The μ' , μ'' , ε' , ε'' values of the samples were calculated with dedicated software.



Scheme 1: Schematic representation of a typical set-up for the measurement of electromagnetic parameters.

III. Results and Discussion

The X-ray diffraction (XRD) pattern of the BaFe₁₂O₁₉ powders synthesized by means of MSS at 1150 °C for 4 h is provided in Fig. 1(a). This pattern shows the formation of a single phase without any traces of unwanted or parasite phases. The diffraction peaks are indexed to JCPDS card no. 43-0002. Most of the peaks of (00h), such as (006), (008) and (0014) are found to have higher intensities than other peaks, indicating the surface of $BaFe_{12}O_{19}$ grains is parallel to the (00h) planes. This suggests that the BaFe₁₂O₁₉ grains have a high degree of grain orientation, and the corresponding SEM image (inset of Fig. 1 (a)) shows that these grains have a plate-like morphology. The growth along the a(b) axis is more pronounced than that along the c-axis, owing to the diffraction peaks of (006), (008) and (0014) planes in the XRD patterns. In Fig. 1(b), the diffraction peaks at 31.8°, 34.6°, and 36.4° originate from ZnO (JCPDS card no. 36-1451). No diffraction peaks of other impurities are detected, which indicates a high purity and crystallinity of these ZnO samples ²⁶. The SEM image (inset of Fig. 1(b)) shows that the synthesized ZnO nanomaterials are globular and of uniform size.

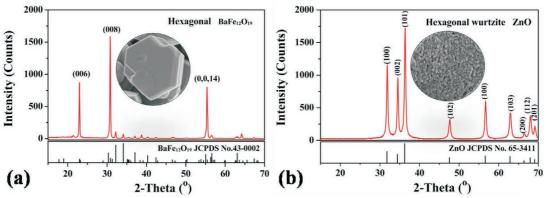


Fig. 1: XRD pattern of the $BaFe_{12}O_{19}$ and ZnO powders. The inset in (a) is a magnified local area of $BaFe_{12}O_{19}$ SEM images and the inset in (b) is an SEM image of ZnO.

The SEM micrographs of the ZnO/BaFe₁₂O₁₉ composite thick films with different mass fraction of BaFe₁₂O₁₉ are presented in Fig. 2. It can be seen that the diameters of ZnO grains in the four thick films coincide well with each other, and are about 50–100 nm, and the thickness and the diameter of BaFe₁₂O₁₉ grains are about 50–70 nm and 10–20 μ m, respectively. Noticeably, BaFe₁₂O₁₉ grains exhibit the high degree of orientation in the films. In order to illustrate the internal structure of ZnO/BaFe₁₂O₁₉ composite thick films, the microstructure and structures of the representative 65 %ZnO/35 %BaFe₁₂O₁₉ composite thick film were investigated with a high-resolution scanning electron microscope (HRSEM), as shown in Fig. 3. From Fig. 3(a) and Fig. 3(b), it can be found that in the 65 %ZnO/35 %BaFe₁₂O₁₉ composite thick film, BaFe₁₂O₁₉ grains are evenly distributed between ZnO grains. From the energy-dispersive spectrometer (EDS) mapping image, it can be easily seen that the O, Zn, Fe and Ba are uniformly distributed, implying that the platelike BaFe₁₂O₁₉ and ZnO spheres are well dispersed in the composites.

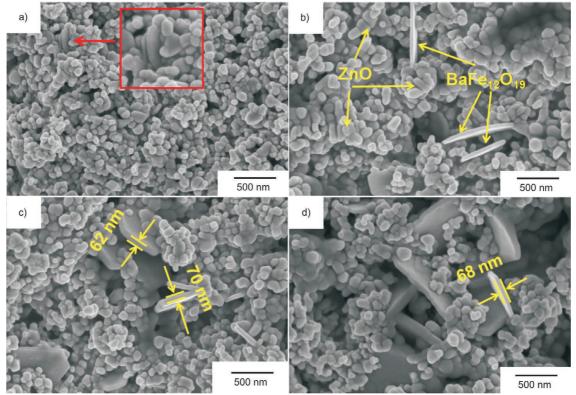


Fig. 2: SEM micrographs of the $(1-x)ZnO/xBaFe_{12}O_{19}$ composite thick films after tape casting: (a) x = 5 wt%, (b) x = 15 wt%, (c) x = 25 wt%, (d) x = 35 wt%.

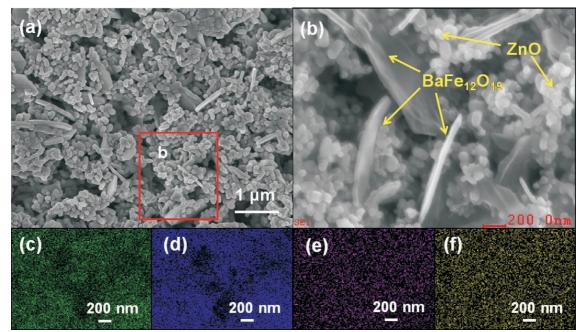


Fig. 3: SEM images of the 65 %ZnO/35 %BaFe₁₂O₁₉ composite thick film at different magnifications (a, b), elemental maps of O (c), Zn (d), Fe (e) and Ba (g) in ZnO/BaFe₁₂O₁₉ composite thick films, respectively.

To verify the magnetic properties of the samples, the hysteresis loop of each sample was measured using VSM, as shown in Fig. 4. It is shown that all the films exhibit a soft magnetic behavior. The saturation magnetization (M_s) of (1-x)ZnO/xBaFe₁₂O₁₉ composite thick films with the values of 6.4, 10.1, 15.5 and 20.7 emu/g have been obtained corresponding to the BaFe₁₂O₁₉ fractions of x = 5, 15, 25 and 35 wt%, respectively. According to Rikukawa ²⁷, for either domain wall movement or spin rotation, the initial permeability is proportional to M_s^2 . As explained above, with increasing BaFe₁₂O₁₉ fraction, M_s increases.

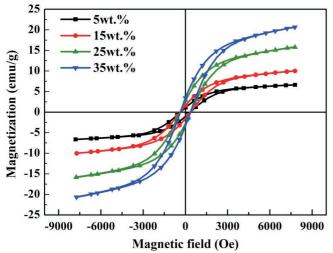


Fig. 4: Magnetization curves of $BaFe_{12}O_{19}/ZnO$ composite thick films with mass fraction of $BaFe_{12}O_{19}$.

The electromagnetic parameters (relative complex permittivity, $\varepsilon_r = \varepsilon' - j\varepsilon''$, and relative complex permeability, $\mu_r = \mu' - j\mu''$) were measured to investigate the microwave absorption properties of the (1-x)ZnO/xBaFe₁₂O₁₉ composite thick films, as shown in Fig. 5. The real permittivity (ε') and real permeability (μ') symbolize the storage ability of electric and magnetic energy, while the imaginary permittivity (ε'') and imaginary permeability (μ'') are related to the dissipation of electric and magnetic energy ^{28, 29}.

As shown in Fig. 5(a), both ε' and ε'' show first an increase but then decrease with the increase of the $BaFe_{12}O_{19}$ fraction in the range of 2-18 GHz. The ε' and ε'' of 75 wt%ZnO/25 wt%BaFe12O19 composite thick film are the highest, in the range of 15.3-17.2 GHz and 12.6-14.5 GHz, respectively. However, too high permittivity is harmful to the impedance match and results in strong reflection and weak absorption ^{30, 31}. In addition, the fluctuation of the ε' and ε'' in the range of 2–18 GHz is ascribed to displacement current lag at the heterogeneous interface. The displacement current lag is caused by the polarization process at the interfaces between ZnO and BaFe₁₂O₁₉ and associated relaxation process ^{32, 33}. The μ' and μ'' are plotted in Fig. 5(b) in the range of 2-18 GHz. As shown in Fig. 5(b), μ' decreases from 2.5 to 0.8 and exhibits broad multi-resonance peaks at 2-18 GHz, implying that natural resonance occurs in the ZnO/BaFe₁₂O₁₉ composite thick films. In addition, it can be speculated that the multi-resonance peaks, are the consequence of the multi-layer interface effect, the surface effect, and spin-wave excitations, defined as "exchange

mode" resonance ^{34–36}. Fig. 5(c) illustrates the dielectric loss of the ZnO/BaFe₁₂O₁₉ composite thick films in the range of 2–18 GHz. It is observed that the dielectric loss first increases and then decreases with the increase of the BaFe₁₂O₁₉ fraction, and the dielectric loss of 75 % ZnO/25 % BaFe₁₂O₁₉ composite thick film is the highest. From the tan δ_{μ} curve in Fig. 5(d), it can be observed that the values of tan δ_{μ} show a decreasing tendency with increasing frequency in the range of 2–18 GHz, besides a slight peak found in the frequency range of 4–6 GHz.

Debye dipolar relaxation can be considered as an important mechanism to account for the dielectric loss of materials. The relative complex permittivity ε_r can be described as ^{37, 38}:

$$\varepsilon_{\rm r} = \varepsilon_{\infty} + \frac{\varepsilon_{\rm s} - \varepsilon_{\infty}}{1 + j2\pi f\tau} = \varepsilon' - j\varepsilon'' \tag{1}$$

where ε_s , ε_∞ , f and τ are the static permittivity, relative dielectric permittivity at the high-frequency limit, frequency and polarization relaxation time, respectively. Thus, ε' and ε'' can be described by

$$\varepsilon' = \varepsilon_{\infty} + \frac{\varepsilon_{s} - \varepsilon_{\infty}}{1 + (2\pi f)^{2} \tau^{2}}$$
(2)

$$\varepsilon'' = \frac{2\pi f \tau (\varepsilon_s - \varepsilon_\infty)}{1 + (2\pi f)^2 \tau^2}$$
(3)

According to Eq. (2) and (3), the relationship between ε' and ε'' can be deduced as

$$\left[\varepsilon' - \frac{\varepsilon_s + \varepsilon_\infty}{2}\right]^2 + (\varepsilon'')^2 = \left[\frac{\varepsilon_2 - \varepsilon_\infty}{2}\right]^2 \tag{4}$$

Thus, the plot of ε' versus ε'' would be a single semicircle, generally denoted as the Cole-Cole semicircle 39. The Cole-Cole plot of the ZnO/BaFe₁₂O₁₉ composite thick films is shown in Fig.6. Each semicircle corresponds to one Debye relaxation process. Three Cole-Cole semicircles can be found for the $ZnO/BaFe_{12}O_{19}$ composite thick films, indicating the existence of a Debye relaxation process. The three relaxation processes of the ZnO/BaFe12O19 composite thick films may arise as follows: under the alternating electromagnetic field, there is a lag of the induced charges from the interfaces of ZnO@BaFe12O19, ZnO@ZnO and BaFe₁₂O₁₉@BaFe₁₂O₁₉, which results in relaxation under the external applied field and transfers the electromagnetic energy to the thermal energy ⁴⁰. Therefore, the microwave energy is attenuated. The Cole-Cole semicircles are distorted, indicating that except for the Debye relaxation, other mechanisms, such as Maxwell-Wagner relaxation and electron polarization, could also exist in the network ZnO/BaFe12O19 composite thick films. In the network structured composites, the additional interfaces can induce interfacial polarizations ^{41, 42}. In addition, the numerous ZnO and BaFe₁₂O₁₉ particles are responsible for interfacial polarization, which further contributes to the dielectric loss. Interfacial polarization occurs in the heterogeneous media due to accumulation of charges at the interfaces.

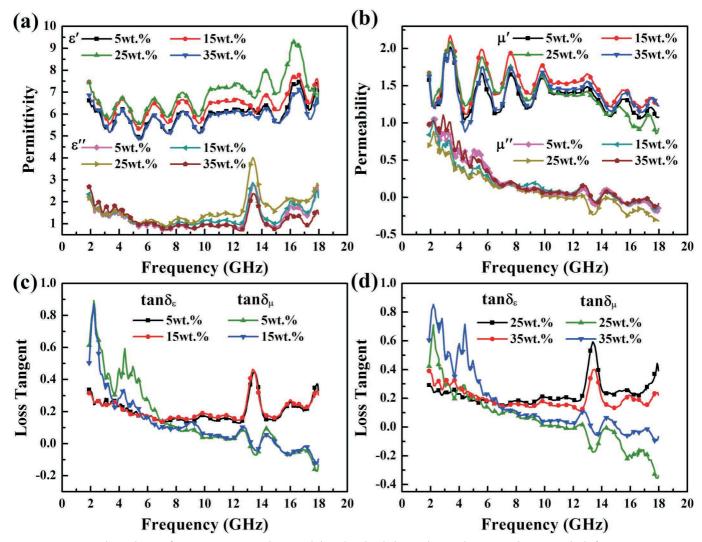


Fig. 5: Frequency dependence of permittivity (a) and permeability (b), the dielectric loss and magnetic loss (c) and (d) for $ZnO/BaFe_{12}O_{19}$ composite thick films with different mass fractions.

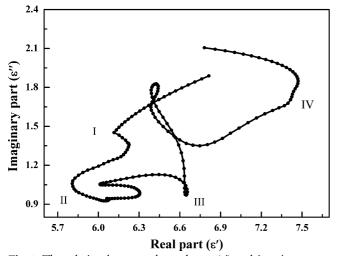


Fig. 6: The relation between the real part (ε') and imaginary part (ε'') of complex permittivity (Cole-Cole plot) of ZnO/BaFe₁₂O₁₉ composite thick films.

It is well known that the magnetic loss is related to domain wall resonance ⁴³, the eddy current effect and the natural resonance of the composite thick films ^{44, 45}. The contribution of domain wall resonance is negligible since it occurs usually in the low frequency range of 1-100 MHz, which is far lower than the measurement frequencies. The eddy current loss is related to the diameter of the particles (d) and electric conductivity (σ), which can be expressed by the following equation ^{46, 47}:

$$\mu'' = 2\pi\mu_0(\mu')^2\sigma d^2 f/3$$
 (5)

where μ_0 is the permeability of vacuum. As a deformation formula:

$$\mu''(\mu^{\text{prime}})^2 f^{-1} = 2\pi\mu_0 \sigma d^2/3 = C_0$$
(6)

If the magnetic loss results from eddy current loss, the value of C_0 is a constant when the frequency is varied ⁴⁸.

Fig. 7 shows the eddy-current loss curves of the ZnO/ BaFe₁₂O₁₉ composite thick films with different mass fractions of BaFe₁₂O₁₉, the C₀ values changes drastically as a function of frequency in the range of 2.0–7.0 GHz. However, it can be seen that the composite thick films have a constant $\mu''(\mu')^{-2}(f)^{-1}$ value from 7.0 to 18.0 GHz, indicating that the eddy-current loss range of the samples is wider in relatively high frequency regions. Besides, the magnetic loss in the ZnO/BaFe₁₂O₁₉ composite thick films is mainly caused by the natural resonance in the frequency range of 2.0–7.0 GHz. For the ZnO/BaFe₁₂O₁₉ composite thick films, the natural resonance can be attributed to

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the small size effect of ZnO. According to the natural-resonance equation ⁴⁹,

$$2\pi f_r = r H_\partial \tag{7}$$

$$H_{\partial} = \frac{4|K_1|}{3\mu_0 M_s} \tag{8}$$

where r is the gyromagnetic ratio, H_a is the anisotropy energy, and $|K_1|$ is the anisotropy coefficient. Generally, the undoped barium hexaferrite has a natural resonance frequency at about 48 GHz due to its strong uniaxial anisotropy along the c-axis ⁵⁰. According to Eq. (8), the anisotropy energy increases with the decrease of saturation magnetization. The M_s value of the ZnO/BaFe₁₂O₁₉ composite thick films increases with the increase of the BaFe₁₂O₁₉ mass fraction (Fig. 4). Hence, the H_∂ of the 65 %ZnO/35 %BaFe₁₂O₁₉ composite thick film is higher. The higher H_∂ is helpful to the improvement of microwave absorption properties ⁵¹.

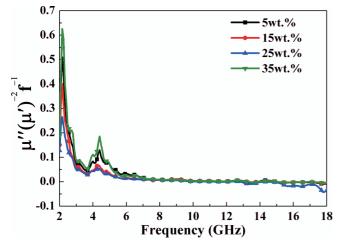


Fig. 7: Frequency dependency of the eddy-current loss curves of $ZnO/BaFe_{12}O_{19}$ composite thick films with different mass fraction of $BaFe_{12}O_{19}$.

In order to avoid the reflection of electromagnetic wave on the surface, the characteristic impedance should be as close as possible to the free space. Z_{im} can be calculated according to the complex permittivity and complex permeability, and the formula is as follows ⁵².

$$Z_{\rm im} = \sqrt{\frac{\mu_0}{\epsilon_0}} \sqrt{\frac{\mu_r}{\epsilon_r}}$$
(9)

Fig. 8 shows that the Z_{im} value of the ZnO/BaFe₁₂O₁₉ composite thick films is 0.30 ~ 0.65 in the whole measurement frequency range, less than Z (Z = 1), which shows that the electromagnetic waves on the sample surface have a larger reflection. According to the impedance matching theory, more electromagnetic waves reflected from the surface of the sample are greatly reduced, and more electromagnetic waves can enter the interior of the material ^{53, 54}. In an alternating electromagnetic field, the macroscopic current would be induced to overcome the variation of the electromagnetic waves. Therefore the electromagnetic waves will be dissipated in the form of Joule heat ⁵⁵.

Reflection loss (RL) represents the microwave absorption property and can be calculated from the relative permeability and permittivity at a given frequency (f) and thickness (d) using the following equations $^{56-58}$:

$$Z_{in} = \sqrt{\frac{u_r}{\varepsilon_r}} \tanh\left(j\frac{2\pi f d}{c}\sqrt{u_r\varepsilon_r}\right)$$
(10)

$$RL(dB) = 20 \lg \left| \frac{Z_{in} - 1}{Z_{in} + 1} \right|$$
(11)

where f is the microwave frequency, d is the thickness of the absorber, c is the velocity of light and Z_{in} is the input impedance of the absorber. The impedance matching condition is determined by the combinations of six parameters (namely, ε' , ε'' , μ' , μ'' , f and d). The reflection loss curve versus frequency can be calculated from ε_r and μ_r at an as-designed layer thickness.

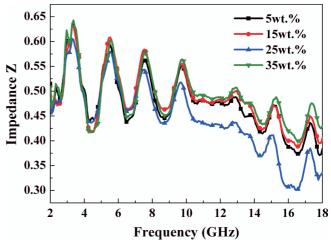


Fig. 8: The impedance of the composites versus frequency of ZnO/ BaFe₁₂O₁₉ composite thick films with different mass fraction of BaFe₁₂O₁₉.

Fig. 9 shows the calculated three-dimensional presentations of theoretical RL of the ZnO/BaFe12O19 composite thick films with different thicknesses in the range of 2–18 GHz. The microwave absorption properties of the ZnO/BaFe₁₂O₁₉ composite thick films and some recently reported ZnO-based composites are shown in Table 1. As shown in Fig. 9(a), the minimum RL reaches -32.4 dB at 12.6 GHz for the $95 \text{ wt}\% \text{ZnO}/5 \text{ wt}\% \text{BaFe}_{12}\text{O}_{19}$ composite thick film with a thickness of 6 mm, and the bandwidth of RL less than -10 dB can reach up to 5.9 GHz. For 85 wt%ZnO/15 wt%BaFe₁₂O₁₉ composite thick film (Fig. 9(b)), the minimum RL reaches -48.6 dB at 17.2 GHz with a thickness of 4.5 mm, whereas for 75 wt%ZnO/25 wt%BaFe12O19 composite thick film (Fig. 9(c)), the minimum RL reaches -36.0 dB at 15.3 GHz with a thickness of 5 mm. For 65 wt%ZnO/35 wt%BaFe₁₂O₁₉ composite thick film (Fig. 9(d)), the minimum RL reaches -36.6 dB at 12.6 GHz with a thickness of 6 mm. From the above data, it can be concluded that the $ZnO/BaFe_{12}O_{19}$ composite thick films have microwave absorption with multiple frequency bands at a certain thickness. The peak at high frequency is related to the natural resonance, and the absorption peak at low frequencies is attributed to the domain wall motion ⁴³. And we can infer that the microwave absorption properties of the ZnO/BaFe12O19 composite thick films can be adjusted based on the thickness and the $BaFe_{12}O_{19}$ mass fraction.

Table 1: The microwave ab	osorption properties of	f ZnO/BaFe ₁₂ O ₁₉ o	composite thick film	ms and some recent	ly reported
composites.					

Samples	Minimum RL value (dB)	Optimum thickness (mm)	Optimum frequency (GHz)	Frequency range (RL<-10dB)	Ref.
Tubular ZnO/CoFe ₂ O ₄ nanocomposites	-28.2 dB	1.5 mm	8.6 GHz	9 GHz	13
BaFe ₁₂ O ₁₉ /ZnO composite	-37.5 dB	6.8 mm	4 GHz	4.3 GHz	2
ZnO-Based materials modified with $ZnAl_2O_4$	-25 dB	2.8mm	10.5 GHz	4.5 GHz	14
Cage-like ZnO/SiO ₂ nanocomposites	-10.6dB	3.0 mm	12.8 GHz	-	18
95 wt%ZnO/5 wt%BaFe ₁₂ O ₁₉	-32.4 dB	6.0 mm	12.6 GHz	5.9 GHz	this work
$85 { m wt\%ZnO/15 wt\%BaFe_{12}O_{19}}$	-48.6 dB	4.5 mm	17.2 GHz	2.3 GHz	this work
75 wt%ZnO/25 wt%BaFe ₁₂ O ₁₉	-36.0 dB	5.0 mm	15.3 GHz	3.8 GHz	this work
$65 \text{ wt}\% \text{ZnO}/35 \text{ wt}\% \text{BaFe}_{12}\text{O}_{19}$	-36.6 dB	6.0 mm	12.6 GHz	2.2 GHz	this work

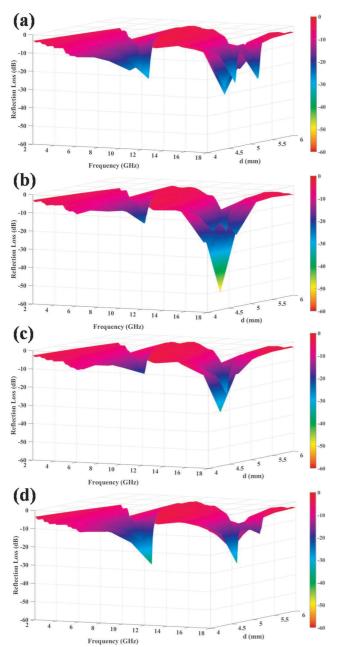


Fig. 9: Three-dimensional and contour plot of the RL versus of the $(1-x)ZnO/xBaFe_{12}O_{19}$ composite thick films (a)x = 5 wt%, (b) 15 wt%, (c) 25 wt%, (d) 35 wt%.

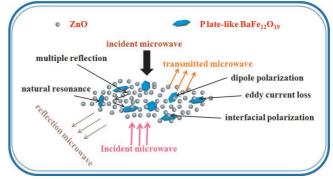


Fig. 10: Schematic illustration of possible EM absorption mechanisms in the $ZnO/BaFe_{12}O_{19}$ composite thick films.

The microwave absorption performances of the ZnO/ BaFe₁₂O₁₉ composite thick films are mainly attributed to two key factors: electromagnetic wave attenuation and impedance matching. To give a visual demonstration of the microwave absorption mechanism, a schematic diagram is presented in Fig. 10. The three-dimensional network will be formed when the plate-like $BaFe_{12}O_{19}$ grains are distributed among the sphere-like ZnO grains. As electromagnetic waves impenetrate the thick film, the energy is induced into dissipative current by the networks, which can be explained by the following facts. Firstly, the multi-interfaces and triple junctions (ZnO@BaFe₁₂O₁₉, ZnO@ZnO, BaFe₁₂O₁₉@BaFe₁₂O₁₉) are advantageous for electromagnetic attenuation owing to the existing interfacial polarizations 59,60. The electromagnetic waves radiating on absorbents are partially absorbed, while the surplus electromagnetic waves will present diffuse reflections owing to the multiple interfaces. During the process of diffuse reflections, the electromagnetic waves enter another process, in which the energy is induced into dissipative current. Secondly, the BaFe₁₂O₁₉ grains and the void space existing between ZnO and BaFe12O19 grains result in relatively large specific surface areas, providing more active sites for reflection and scattering of electromagnetic waves ⁶¹. Finally, the void space between ZnO and BaFe₁₂O₁₉ grains can effectively interrupt the spread of electromagnetic waves and generate dissipation owing to the existing impendence difference and enhance the microwave absorption properties 62 . The results could provide guidance for exploring and designing advanced microwave absorption materials based on adjustment of the BaFe₁₂O₁₉ mass fraction and the thickness of the composite thick films.

IV. Conclusions

In conclusion, ZnO/BaFe₁₂O₁₉ composite thick films have been successfully synthesized with the tape-casting process with the presence of synthesized plate-like BaFe₁₂O₁₉ grains and ZnO sphere-like grains, in which interweaves with ZnO spheres and a three-dimensional network can be formed. The phase composition, morphology and magnetic properties of the ZnO/BaFe₁₂O₁₉ composite thick films are also characterized and discussed. The electromagnetic data demonstrate that the microwave absorption property of the thick films can be adjusted easily by varying the mass ratio of BaFe₁₂O₁₉. When the mass ratio of BaFe₁₂O₁₉ is 15, the composite thick film exhibits a minimum RL of -48.6 dB at 17.2 GHz with a thickness of 4.5 mm. Such composite thick films could be developed for a wide spectrum of applications in the area of microwave absorption.

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